New Quaternary Indides $RE_7 \text{Ni}_{5-x} \text{Ge}_{3+x} \text{In}_6 \ (RE = \text{La, Nd, Sm})$

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The quaternary indides $RE_7\mathrm{Ni}_{5-x}\mathrm{Ge}_{3+x}\mathrm{In}_6$ ($RE=\mathrm{La},\mathrm{Nd},\mathrm{Sm}$) were synthesized from the elements by arc-melting. Single crystals were grown by slow cooling of the polycrystalline samples. The structures were characterized by powder and single-crystal X-ray diffraction: $\mathrm{Ce}_7\mathrm{Ni}_{4.73}\mathrm{Ge}_{3.27}\mathrm{In}_6$ type, P6/m, Z=1, a=1147.05(9), c=426.82(4) pm, wR2=0.0652, 528 F^2 values for $\mathrm{La}_7\mathrm{Ni}_{4.46}\mathrm{Ge}_{3.54}\mathrm{In}_6$, a=1134.5(7), c=407.1(7) pm, wR2=0.0419, 441 F^2 values for $\mathrm{Nd}_7\mathrm{Ni}_{4.91}\mathrm{Ge}_{3.09}\mathrm{In}_6$, and a=1133.5(2), c=404.3(1) pm, wR2=0.0619, 498 F^2 values for $\mathrm{Sm}_7\mathrm{Ni}_{4.31}\mathrm{Ge}_{3.69}\mathrm{In}_6$, with 25 parameters per refinement. Characteristic features of the $RE_7\mathrm{Ni}_{5-x}\mathrm{Ge}_{3+x}\mathrm{In}_6$ structures are hexagonal, AlB_2 -related prisms around the RE1 atoms and a tricapped, trigonal-prismatic coordination of the nickel atoms.

Key words: Indide, Rare Earth Compounds, Crystal Structure

Introduction

Several structure types of binary intermetallic compounds RE_xM_y (RE = rare earth element; M = element of the 3rd, 4th, or 5th main group) have ternary and quaternary representatives formed by an ordered occupancy of the RE or M positions by two different elements. Typical examples are the series RE_{11} Ge₄In₆ with an ordered Ho₁₁Ge₁₀-type structure [1,2], RE_3 GeIn with an ordered Gd₃Ga₂-type [3,4], RE_2 Ge₂In with an ordered U₃Si₂-type [5 – 9], RE_4 Ni₂InGe₄ with an ordered Mg₅Si₆-type [10], and Yb₃AuGe₂In₃ with an ordered ZrNiAl-type structure [11].

Besides these ordered versions of binaries, also some individual structure types of ternary and quaternary indium germanides exist. SrInGe and EuInGe [12] are ternary Zintl phases with an anionic network derived from the ThSi₂ type. AE_2 LiInGe₂ (AE = Ca, Sr) [13] and SrCa₂In₂Ge [14] contain chain-like polyanions which are separated and charge balanced by the alkali and alkaline earth metal cations. A more complex three-dimensional [Co₄InGe₁₂] network with corner-sharing CoGe₅ square pyramids occurs for the series RE_7 Co₄InGe₁₂ (RE = Dy, Ho, Yb) [15]. Yb₇Co₄InGe₁₂ exhibits mixed ytterbium valency.

Recently we reported on the quaternary compounds $Ce_7Ni_{5\pm x}Ge_{3\pm x}In_6$ and $Pr_7Ni_{5\pm x}Ge_{3\pm x}In_6$ [16]. The latter crystallize with a new structure type which derives from the AlB_2 type by substitution of the aluminum and boron sites by $RE(Ni/Ge)_{12}$ and $NiIn_6RE_3$ building groups. During our phase-analytical studies of quaternary RE–Ni–Ge–In systems we obtained the isotypic compounds $RE_7Ni_{5-x}Ge_{3+x}In_6$ (RE = La, Nd, Sm) which are reported herein.

Experimental Section

Synthesis

Starting materials for the synthesis of the RE₇Ni_{5-x}-Ge3+xIn6 samples were pieces of the rare earth elements (smart elements or Kelpin, > 99.9 %), nickel wire (Johnson Matthey, Ø 1 mm), germanium lumps (Wacker), and indium tear drops (Chempur), all with stated purities better than 99.9 %. Pieces of the rare earth metals were first arc-melted [17] to small buttons under an argon pressure of ca. 700 mbar. The argon was purified with titanium sponge (900 K), silica gel, and molecular sieves. The four elements were then weighed in the ideal 7:5:3:6 ratio and arcmelted. To ensure homogeneity of the samples, the buttons were turned over and remelted twice. In all cases the total weight loss was smaller than 0.5 %. The $RE_7Ni_{5-x}Ge_{3+x}In_6$ samples were obtained as polycrystalline powders after this heat treatment. For crystal growth, parts of these samples were placed in tantalum crucibles which were sealed into evacuated silica ampoules. The samples were then heated to 1280 K within 5 h and kept at that temperature for another 5 h. The temperature was then decreased at a rate of 5 K h^{-1} to 1000 K, and to 700 K at a rate of 15 K h^{-1} . Finally the samples were cooled to r.t. by switching off the furnace. These heat treatments resulted in brittle, air-stable samples with small single crystals which exhibited metallic

Refined composition	La ₇ Ni _{4,46} Ge _{3,54} In ₆	Nd7Ni4.91Ge3.09In6	Sm ₇ Ni _{4.31} Ge _{3.69} In ₆
Formula weight, g mol ⁻¹	2180.11	2211.17	2262.27
Unit cell dimensions (Guinier data)			
a, pm	1147.05(9)	1134.5(7)	1133.5(2)
c, pm	426.82(4)	407.1(7)	404.3(1)
Cell volume, nm ³	0.4863	0.4538	0.4499
Calculated density, g cm ⁻³	7.44	8.09	8.35
<i>F</i> (000), e	931	950	967
Crystal size, μ m ³	$10 \times 10 \times 30$	$15 \times 15 \times 20$	$10 \times 15 \times 35$
Transm. ratio (max / min)	0.645 / 0.263	0.287 / 0.233	0.341 / 0.206
Absorption coefficient, mm ^{−1}	31.5	37.0	40.3
Detector distance, mm	80	_	_
Exposure time, min	3	_	-
ω range; increment, deg	0-180, 1.0	_	_
Integr. param. A, B, EMS	12.6, 1.8, 0.018	_	-
θ range for data collection, deg	2.0 - 30.0	2.0 - 28.5	2.0 - 30.0
Range in hkl	$\pm 16, \pm 16, \pm 6$	$\pm 15, \pm 15, \pm 5$	$\pm 15, \pm 15, \pm 5$
Total no. reflections	4257	4444	5230
Independent reflections / R_{int}	528 / 0.0822	441 / 0.2062	498 / 0.1897
Reflections with $I \ge 2\sigma(I) / R_{\sigma}$	380 / 0.0645	305 / 0.0683	345 / 0.0589
Data / ref. parameters	528 / 25	441 / 25	498 / 25
$R1 / wR2$ for $I \ge 2\sigma(I)$	0.0311 / 0.0624	0.0327 / 0.0667	0.0312 / 0.0680
R1 / wR2 for all data	0.0508 / 0.0562	0.0359 / 0.0419	0.0528 / 0.0619
Extinction coefficient	0.0058(3)	0.00111(6)	0.00025(7)
Goodness-of-fit on F^2	0.885	1.033	1.057
Largest diff. peak / hole, e Å ⁻³	2.03 / -2.16	3.81 / -2.77	2.50 / -2.54

Table 1. Crystal data and structure refinement for $RE_7\mathrm{Ni}_{5-x}$ - $\mathrm{Ge}_{3+x}\mathrm{In}_6$ ($RE=\mathrm{La},\mathrm{Nd},\mathrm{Sm}$), space group P6/m,Z=1.

EDX data

The three single crystals investigated on the diffractometers were studied by EDX using a Zeiss EVO MA10 scanning electron microscope with LaB₆, NdF₃, SmF₃, Ni, Ge, and InAs as standards for the semiquantitative measurements. The results of the analyses were in agreement with the compositions refined from the single crystal X-ray data. No impurity elements were observed.

X-Ray diffraction

The polycrystalline $RE_7\mathrm{Ni}_{5-x}\mathrm{Ge}_{3+x}\mathrm{In}_6$ samples were studied through Guinier patterns (image plate system Fujifilm, BAS-1800) using $\mathrm{Cu}K_{\alpha 1}$ radiation and α -quartz (a=491.30, c=540.46 pm) as an internal standard. The hexagonal lattice parameters were deduced from least-squares refinements. Correct indexing of the patterns was ensured by intensity calculations [18] using the positional parameters obtained from the structure refinements.

Irregularly shaped crystal fragments of $RE_7\mathrm{Ni}_{5-x^-}$ Ge_{3+x}In₆ were separated from the annealed samples by mechanical fragmentation. The crystals were glued to quartz fibers using beeswax and characterized by Laue photographs (Buerger camera, white molybdenum radiation, image plate technique, Fuji-film, BAS-1800) in order to check their suitability for intensity data collections. The data sets of Nd₇Ni_{4.91}Ge_{3.09}In₆ and Sm₇Ni_{4.31}Ge_{3.69}In₆ were collected at r.t. by use of a four-circle diffractometer (CAD4) with graphite-monochromatized Mo K_{α} radiation and a scintilla-

tion counter with pulse height discrimination. Scans were taken in the $\omega/2\theta$ mode. Empirical absorption corrections were applied on the basis of Ψ -scan data, accompanied by spherical absorption corrections. The La₇Ni_{4.46}Ge_{3.54}In₆ crystal was measured at r.t. by use of an IPDS II diffractometer (graphite-monochromatized Mo K_{α} radiation; oscillation mode). A numerical absorption correction was applied to the data set. All relevant crystallographic data and details of the data collections and evaluations are listed in Table 1.

Structure refinements

The isotypy of the $RE_7\mathrm{Ni}_{5-x}\mathrm{Ge}_{3+x}\mathrm{In}_6$ indides with $\mathrm{Ce}_7\mathrm{Ni}_{4.73}\mathrm{Ge}_{3.27}\mathrm{In}_6$ [16] was evident from the Guinier patterns. The three data sets showed low hexagonal Laue symmetry and no further systematic extinctions, in agreement with space group P6/m. The atomic positions of $\mathrm{Ce}_7\mathrm{Ni}_{4.73}\mathrm{Ge}_{3.27}\mathrm{In}_6$ [16] were taken as starting values, and the structures were refined with anisotropic displacement parameters for all atoms with SHELXL-97 (full-matrix least-squares on $F_0{}^2$) [19]. Similar to the cerium and praseodymium compound [16], the 6k sites showed mixed Ni/Ge (M) occupancy. These occupancies were refined as least-squares variables, leading to the compositions listed in Table 1. The atomic parameters and interatomic distances are listed in Tables 2 and 3.

Further details of the crystal structure investigation may be obtained from Fachinformationszentrum Karlsruhe, 76344 Eggenstein-Leopoldshafen, Germany (fax: +49-7247-808-666; e-mail: crysdata@fiz-karlsruhe.de, http://www.fiz-

Table 2. Atomic coordinates and anisotropic displacement parameters (pm²) for $RE_7Ni_{5-x}Ge_{3+x}In_6$ (RE = La, Nd, Sm). U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor. The anisotropic displacement factor exponent takes the form: $-2\pi^2[(ha^*)2U_{11} + ... + 2hka^*b^*U_{12}]$. $U_{23} = U_{13} = 0$.

Atom	Wyck. site	х	у	z	U_{11}	U_{22}	U_{33}	U_{12}	$U_{\rm eq}$
La ₇ Ni _{4.46} Ge _{3.54} In ₆									
La1	1 <i>a</i>	0	0	0	79(5)	U_{11}	117(9)	40(3)	92(4)
La2	6j	0.30466(7)	0.38123(7)	0	103(4)	79(3)	103(3)	39(3)	98(2)
In	6 <i>k</i>	0.16972(9)	0.51796(9)	1/2	101(5)	74(5)	131(4)	32(4)	107(2)
Ni2	2c	1/3	2/3	0	148(9)	U_{11}	107(14)	74(5)	135(6)
M^{a}	6 <i>k</i>	0.2397(2)	0.1571(2)	1/2	142(10)	106(9)	143(9)	79(7)	123(7)
$Nd_{7}Ni_{4.91}Ge_{3.09}In_{6}$									
Nd1	1 <i>a</i>	0	0	0	69(6)	U_{11}	73(9)	35(3)	70(4)
Nd2	6 <i>j</i>	0.30003(8)	0.37968(8)	0	108(4)	75(4)	80(3)	40(3)	90(2)
In	6 <i>k</i>	0.16350(11)	0.51651(11)	1/2	99(5)	89(5)	98(5)	38(5)	99(2)
Ni2	2c	1/3	2/3	0	145(10)	U_{11}	81(15)	73(5)	124(7)
M^{b}	6 <i>k</i>	0.2392(2)	0.1599(2)	1/2	143(11)	84(11)	103(9)	69(8)	104(7)
$Sm_7Ni_{4.31}Ge_{3.69}In_6$									
Sm1	1 <i>a</i>	0	0	0	82(5)	U_{11}	72(7)	41(3)	79(4)
Sm2	6 <i>j</i>	0.29795(8)	0.37921(8)	0	116(4)	82(3)	54(3)	39(3)	89(2)
In	6 <i>k</i>	0.16082(11)	0.51572(11)	1/2	112(5)	97(5)	89(4)	52(4)	99(2)
Ni2	2c	1/3	2/3	0	136(10)	U_{11}	76(13)	68(5)	116(6)
M ^c	6 <i>k</i>	0.2387(2)	0.1611(2)	1/2	166(11)	114(10)	97(9)	101(8)	112(7)

^a M = 0.59(7) Ge + 0.41(7) Ni; ^b M = 0.51(7) Ge + 0.49(7) Ni; ^c M = 0.62(7) Ge + 0.38(7) Ni.

Table 3. Interatomic distances (pm) in La₇Ni_{4.46}Ge_{3.54}In₆. All distances within the first coordination spheres are listed.

La1:	12 M	322.6(1)	In:	2 Ni2	279.0(1)
	6 La2	400.7(1)		1 M	282.8(2)
La2:	2 M	308.9(1)		2 In	311.3(2)
	1 Ni2	312.3(1)		2 La2	340.7(1)
	2 M	313.1(1)		2 La2	344.1(1)
	2 In	340.7(1)		2 La2	346.6(1)
	2 In	344.1(1)		1 In	370.5(2)
	2 In	346.6(1)	Ni2:	6 In	279.0(1)
	1 La2	391.1(1)		3 La2	312.3(1)
	3 La2	400.7(1)	M:	2 M	241.9(1)
				1 In	282.8(2)
				2 La2	308.9(1)
				2 La2	313.1(1)
				2 La1	322.6(1)

informations dienste.de/en/DB/icsd/depot_anforderung.html) on quoting the deposition numbers CSD-422612 (La₇Ni_{4.46}-Ge_{3.54}In₆), CSD-422613 (Nd₇Ni_{4.91}Ge_{3.09}In₆), and CSD-422614 (Sm₇Ni_{4.31}Ge_{3.69}In₆).

Discussion

The intermetallic compounds $RE_7 Ni_{5-x} Ge_{3+x} In_6$ (RE = La, Nd, Sm) are isotypic with the recently reported phases $Ce_7 Ni_{5\pm x} Ge_{3\pm x} In_6$ and $Pr_7 Ni_{5\pm x} Ge_{3\pm x} In_6$ [16]. All five compounds show small homogeneity ranges due to Ni/Ge mixing on the 6k site. No hints for Ni/Ge ordering were evident from the single-crystal X-ray data.

A projection of the La₇Ni_{4.46}Ge_{3.54}In₆ structure is exemplarily given in Fig. 1. The indium and germa-

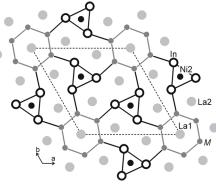


Fig. 1. Projection of the La₇Ni_{4.46}Ge_{3.54}In₆ structure onto the xy plane. Lanthanum, nickel, germanium, and indium atoms are drawn as light-grey, black filled, small medium-grey, and black open circles, respectively. All atoms lie on mirror planes at z=0 (La1, La2, Ni2) and z=1/2 (M, In). The hexagonal-prismatic and trigonal-prismatic coordinations of La1 and Ni2, respectively, are emphasized.

nium atoms build up layers at z = 1/2 which leave trigonal prismatic voids for the Ni2, distorted pentagonal prismatic voids for the La2, and hexagonal prismatic voids for the La1 atoms. These three types of prisms are capped by three, five, and six atoms on the rectangular faces, leading to coordination numbers 9, 15, and 18, respectively, as frequently observed for related intermetallic phases [20].

The M-M distances of 242 pm within the M_6 hexagons are slightly longer than the sum of the covalent radii of 237 pm (Ni + Ge) [21], but shorter

than in fcc nickel (249 pm) and elemental germanium (245 pm) [22]. We can therefore assume substantial Ni–Ge bonding within the hexagons. The triangular edges of the Ni2In₆ prisms have In–In distances of 311 pm. These distances are even shorter than in elemental tetragonal body-centered indium. There, each indium atom has four nearest indium neighbors at 325 pm and eight further neighbors at 338 pm [22]. The planar M_6 In₆ layers are condensed to a three-dimensional network via the Ni2 atoms within the tri-

gonal prisms. The La1 and La2 atoms are connected to the network via weaker La1–M (323 pm) and La2–M (309 pm) bonds. These distances are longer than the sums of the covalent radii [21] of 284 pm (La + Ni) and 291 pm (La + Ge).

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